



Thiols in the Interstellar Medium: First Detection of HC(O)SH and Confirmation of C₂H₅SH

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Abstract

The chemical compounds carrying the thiol group (-SH) have been considered essential in recent prebiotic studies regarding the polymerization of amino acids. We have searched for this kind of compound toward the Galactic Center quiescent cloud G+0.693–0.027. We report the first detection in the interstellar space of the trans-isomer of monothioformic acid (t-HC(O)SH) with an abundance of $\sim 1 \times 10^{-10}$. Additionally, we provide a solid confirmation of the gauche isomer of ethyl mercaptan (g-C₂H₅SH) with an abundance of $\sim 3 \times 10^{-10}$, and we also detect methyl mercaptan (CH₃SH) with an abundance of $\sim 5 \times 10^{-9}$. Abundance ratios were calculated for the three SH-bearing species and their OH analogs, revealing similar trends between alcohols and thiols with increasing complexity. Possible chemical routes for the interstellar synthesis of t-HC(O)SH, CH₃SH, and C₂H₅SH are discussed, as well as the relevance of these compounds in the synthesis of prebiotic proteins in the primitive Earth.

Unified Astronomy Thesaurus concepts: [Astrochemistry \(75\)](#); [Chemical abundances \(224\)](#); [Interstellar molecules \(849\)](#); [Galactic center \(565\)](#)

1. Introduction

Among different theories of origin of life, one recurrent conundrum is the abiotic polymerization of amino acids since it requires ribosomes, macromolecular machines containing ribonucleic acid (RNA) and proteins. How could the first proteins form if they were needed to synthesize others? Following the ideas of Foden et al. (2020), a possible solution involves a thiol-based scenario in which SH-bearing molecules, together with the family of thioacids (R-C(O)SH) and thioesters (R-S-R'), have important properties as energy carriers and catalysts (Chandru et al. 2016; Leman & Ghadiri 2017). Although these types of compounds could be created in situ by an H₂S-mediated chemistry under prebiotically plausible conditions on early Earth (Shalayel et al. 2020), they also could have been delivered exogenously. Hence, observations of thiol-based molecules in space could shed some light on the availability of such compounds on a primitive Earth, and on their role in the prebiotic synthesis of proteins.

More than 220 molecules have been detected in the interstellar medium (ISM) and circumstellar shells⁹ to date. However, sulfur-containing species only account for 20 of them. Furthermore, while molecules detected carrying C, H, or N range from 2 up to 13 atoms, the vast majority of S-bearing molecules have, at most, 4 atoms (such as H₂CS; Sinclair et al. 1973). This could be due to the relatively low cosmic abundance of atomic sulfur ($\approx 10^{-5}$ with respect to H₂, that is, more than 10 times lower than C or O; Asplund et al. 2009) together with its ability to have many different oxidation states

and allotropes when compared to the more abundant elements (Jiménez-Escobar et al. 2014; Shingledecker et al. 2020) and its capacity of depleting fast in dense molecular clouds (Laas & Caselli 2019).

As a consequence, very few S-bearing molecules containing more than four atoms have firmly been detected in the ISM so far. One example is methyl mercaptan (hereafter CH₃SH), which has been detected in several environments, such as prestellar cores (Gibb et al. 2000), massive star-forming regions like Sagittarius B2 (Linke et al. 1979; Müller et al. 2016), and Solar-like protostars (Majumdar et al. 2016). The other one is ethyl mercaptan (hereafter C₂H₅SH), which was tentatively detected toward Orion KL (Kolesniková et al. 2014). Other searches for complex S-bearing molecules were unsuccessful, such as CH₃CHS (thioacetaldehyde; Margulès et al. 2020), NH₂CHS (thioformamide; Motiyenko et al. 2020), and CH₃SC(O)H (S-methyl thioformate; Jabri et al. 2020).

In this Letter, we report the first detection in the ISM of the trans-isomer of monothioformic acid (hereafter HC(O)SH), the simplest thiol acid. We also report a solid confirmation of the gauche isomer of C₂H₅SH, together with the detection of CH₃SH. These molecules are found toward the quiescent Giant Molecular Cloud G+0.693–0.027 located in the Galactic Center (hereafter G+0.693). This source shows a very rich chemistry with up to 40 different complex organic molecules¹⁰ detected (see, e.g., Requena-Torres et al. 2008; Rivilla et al. 2018, 2019; Zeng et al. 2018; Rivilla et al. 2020; Jiménez-Serra et al. 2020). Studies suggest that this cloud could be undergoing a cloud–

⁹ <https://cdms.astro.uni-koeln.de/classic/molecules>

¹⁰ These are usually referred as carbon-based molecules that have six or more atoms (Herbst & Van Dishoeck 2009).

cloud collision (Zeng et al. 2020), which induces large-scale shocks that sputter dust grains and that enhances the gas-phase abundance of molecules by several orders of magnitude (Requena-Torres et al. 2006).

2. Observations

We have used a spectral line survey toward G+0.693 covering several windows between 32 GHz to 172 GHz with an average resolution of 1.5 km s^{-1} , although the final spectral resolution employed in the figures has been smoothed up to 3 km s^{-1} . We stress that this is for a proper line visualization and does not alter in any form the analysis done.

For the observations, we used both the IRAM 30 m telescope located at Pico Veleta (Granada, Spain) and the Yebes 40 m telescope¹¹ (Guadalajara, Spain). The equatorial coordinates of the molecular cloud G+0.693 are $\alpha(\text{J2000.0}) = 17^{\text{h}}47^{\text{m}}22^{\text{s}}$ and $\delta(\text{J2000.0}) = -28^{\circ}21'27''$. The position switching mode was used in all the observations with the off position located at $(-885'', 290'')$ from the source. The line intensity of the spectra was measured in units of T_{A}^* as the molecular emission toward G+0.693 is extended over the beam (Requena-Torres et al. 2006; Martín et al. 2008; Rivilla et al. 2018). In all the observations, each frequency setup was repeated shifting the central frequency by 20–100 MHz in order to identify spurious lines or contamination from the image band.

The IRAM 30 m observations were performed during three different sessions in 2019: April 10–16, August 13–19, and December 11–15. The dual polarization receiver EMIR was used connected to the fast Fourier transform spectrometers (FFTSs), which provided a channel width of 200 kHz in the 3 and 2 mm radio windows. The observations with the Yebes 40 m radiotelescope were carried out in 2020: from February 3 to 9 and from February 15 to 22. In this case, the Nanocosmos Q-band (7 mm) HEMT receiver was used, which enables ultra-broadband observations in two linear polarizations (Tercero et al. 2021). The receiver was connected to 16 FFTSs providing a channel width of 38 kHz and an instantaneous bandwidth of 18.5 GHz per polarization, covering the frequency range between 31.3 GHz and 50.6 GHz.

3. Analysis and Results

We used the software MADCUBA¹² (Martín et al. 2019) to perform the data analysis and line identification. The Spectral Line Identification and Modelling (SLIM) tool of MADCUBA uses the spectroscopic data entries from different molecular catalogs, and generates a synthetic spectra based under the assumption of local thermodynamic equilibrium (LTE) conditions, and considering line opacity effects. The fitted parameters used to reproduce the molecular emission are column density (N), excitation temperature (T_{ex}), local standard of rest velocity (v_{LSR}), and FWHM.

Both HC(O)SH and C₂H₅SH have two rotamers, associated with the rotation of their respective C-S bond. HC(O)SH has one cis (c-HC(O)SH) and one trans (t-HC(O)SH) rotamer, the former being $\sim 330 \text{ K}$ higher in energy (Hocking & Winnewisser 1976). C₂H₅SH has two degenerated \pm gauche isomers (g-C₂H₅SH) and one antiperiplanar.

For our analysis we have used the following spectroscopic entries from the the CDMS¹³ catalog: entries 062515/062516 for trans/cis isomers of HC(O)SH (Hocking & Winnewisser 1976), 062523/062524 for gauche/anti isomers of C₂H₅SH (Kolesníková et al. 2014; Müller et al. 2016), and 048510 for CH₃SH (Xu et al. 2012; Zakharenko et al. 2019).

3.1. Detection of t-HC(O)SH

The fitting procedure was performed considering the total emission of any other identified molecule in the spectral survey. This evaluation was carried out introducing all the compounds already detected in the ISM and in G+0.693 (see Requena-Torres et al. 2006, 2008; Zeng et al. 2018; Rivilla et al. 2019; Jiménez-Serra et al. 2020). For the analysis, it was also assumed a cosmic microwave background temperature (T_{bg}) of 2.73 K and no background continuum source (Zeng et al. 2020).

The global fit of all rotational lines to the observed data is shown in blue lines in Figure 1, while in red we show the fit of the individual lines of t-HC(O)SH, and the observational data in black. As shown in both Figure 1 and Table 1, we have detected a total of nine a-type transitions in the 7 mm and 3 mm bands, each one of them detected above the 5σ level in integrated intensity. Note that the SLIM synthetic spectrum shows small deviations with respect to the observations. These deviations are larger than the ones obtained for CH₃SH and g-C₂H₅SH (see Figures 3 and 4 in Appendix A). This is due to the fact that the t-HC(O)SH lines are significantly weaker than those of CH₃SH and g-C₂H₅SH (3–7 mK versus 5–12 mK and $>40 \text{ mK}$, respectively). However, the fit of all clean t-HC(O)SH lines is consistent with the noise. We also note that for the three slightly blended lines of t-HC(O)SH, the contribution of the other molecules is less than 10% of the total intensity. Despite that the remaining two appear blended, the predicted LTE intensities are consistent with the observed lines. Note that the rest of the t-HC(O)SH lines covered within the observed frequency range are not shown due to strong blending issues.

The physical parameters obtained from the fit are listed in Table 2. The values obtained for T_{ex} and FWHM are consistent with those obtained previously for other molecular species toward this cloud ($T_{\text{ex}} \sim 5\text{--}20 \text{ K}$ and line widths around 20 km s^{-1} ; Zeng et al. 2018; Rivilla et al. 2020). The low T_{ex} indicates that the emission of the molecules in G+0.693 is subthermally excited as a result of the low H₂ densities of this source ($T_{\text{kin}} \sim 150 \text{ K}$; Requena-Torres et al. 2006; Zeng et al. 2018).

For t-HC(O)SH, the fitted column density gives $(1.6 \pm 0.1) \times 10^{13} \text{ cm}^{-2}$. In addition, we derived a 3σ upper limit of $\leq 3 \times 10^{12} \text{ cm}^{-2}$ for c-HC(O)SH (Table 2), as no clear transition has been detected within our data set. For the calculation, we have assumed the same T_{ex} , v_{LSR} , and FWHM as for t-HC(O)SH. A ratio c-HC(O)SH/t-HC(O)SH ≤ 0.2 was obtained.

3.2. Detection of g-C₂H₅SH and CH₃SH

This molecule was tentatively detected in Orion KL (Kolesníková et al. 2014), based on a few isolated transitions assigned to this isomer. We report here an unambiguous detection of this isomer that confirms its presence in the ISM. We have listed in Table 4 (Appendix A) the transitions measured toward G+0.693. Note that eight of the targeted lines

¹¹ Yebes Observatory is operated by the Spanish Geographic Institute (IGN, Ministerio de Transportes, Movilidad y Agencia Urbana).

¹² MADrid Data CUBe Analysis is a software developed at the Center of Astrobiology in Madrid: <https://cab.inta-csic.es/madcuba/Portada.html>.

¹³ Cologne Database for Molecular Spectroscopy (Endres et al. 2016), <https://cdms.astro.uni-koeln.de/classic/>.

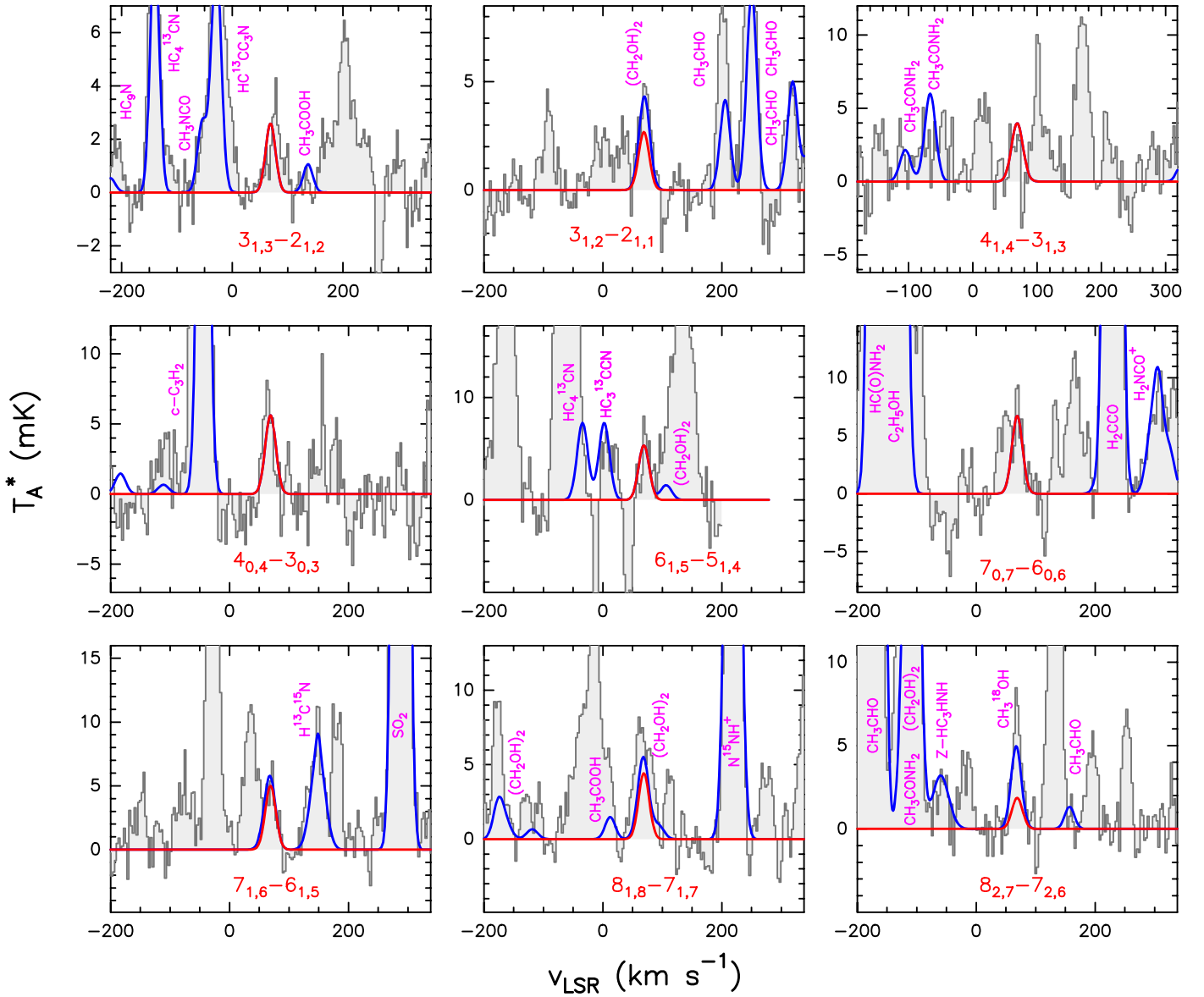


Figure 1. Cleanest and brightest lines of t-HC(O)SH detected toward G+0.693 labeled with their corresponding quantum numbers in red. The red line shows the best LTE fit to the observed spectra (represented by the black lines). The data have been smoothed up to 3 km s^{-1} for an optimal line visualization. The blue lines show the spectra including the emission of all the molecules searched toward G+0.693. Note that these lines are tagged with their corresponding molecular compound in pink.

are totally clean (Figure 3, Appendix A) and above the 5σ level in integrated intensity. We derived a total column density of $(4 \pm 2) \times 10^{13} \text{ cm}^{-2}$ (Table 2).

In the case of CH_3SH , the fit was carried out separating the targeted lines into its $K_a = 0$ and $K_a = 1$ levels¹⁴. The brightest transitions fall into the 3 mm and 2 mm bands as presented in Figure 4 and Table 5 (Appendix B). Note that the agreement between the predicted and observed spectra is excellent for all clean transitions. The column density using the $K_a = 0$ and $K_a = 1$ levels is $(6.5 \pm 0.2) \times 10^{14} \text{ cm}^{-2}$ (Table 2). This gives a ratio of $\text{CH}_3\text{SH}/\text{g-C}_2\text{H}_5\text{SH} = 16 \pm 7$. Note that the column density of $K_a = 2$ and remaining transitions in CH_3SH contribute less than a 10% in the total determined.

Finally, we note that both ^{13}C and ^{34}S isotopologues of CH_3SH have also been detected with a few clean lines and will

be presented in a forthcoming paper (L. Colzi et al. 2021, in preparation).

3.3. Molecular Abundances and Comparison with Their O-bearing Analogs

To derive the fractional abundances of these species relative to H_2 , we have assumed a H_2 column density of $N_{\text{H}_2} = 1.35 \times 10^{23} \text{ cm}^{-2}$ (Martín et al. 2008). This gives $\sim 3 \times 10^{-10}$, $\sim 1 \times 10^{-10}$, and $\sim 5 \times 10^{-9}$ for g- $\text{C}_2\text{H}_5\text{SH}$, t-HC(O)SH, and CH_3SH , respectively (Table 2).

In Figure 2 we have plotted these values and compared them with the abundances obtained for their OH molecular analogs, namely, $\text{C}_2\text{H}_5\text{OH}$ (ethanol), HC(O)OH (formic acid), and CH_3OH (methanol). We obtained a ratio $\text{CH}_3\text{OH}/\text{CH}_3\text{SH} = 23$, $\text{C}_2\text{H}_5\text{OH}/\text{C}_2\text{H}_5\text{SH} = 15$, and $\text{HC(O)OH}/\text{HC(O)SH} = 13$. Although these OH analogs are more abundant by a factor ≥ 10 , the resulting trend is strikingly similar. Note, however, that this trend is lost when we compare molecules such as carbon

¹⁴ The moments of inertia of CH_3SH frame it in the limiting prolate case, $\kappa = (2B - A - C)/(A - C) \approx -0.988$.

Table 1

Lines of t-HC(O)SH Detected toward G+0.693 with Their Corresponding Quantum Numbers (QNs), Logarithm of the Einstein Coefficients ($\log A_{ul}$), Degeneracy (g_u), and Energy (E_u) of the Upper State

Rest Frequency (MHz)	QNs ^a	g_u	E_u (K)	$\log A_{ul}$ (s^{-1})	rms (mK)	$\int T_A^* d\nu^b$ (mK km s ⁻¹)	S/N ^b	Comments
34248.82	3 _{1,3} → 2 _{1,2}	7	4.3	-6.4784	1.4	56	7.3	clean transition
35915.48	3 _{1,2} → 2 _{1,1}	7	4.4	-6.4164	1.4	58	7.6	blended with HOCH ₂ CH ₂ OH
45659.99	4 _{1,4} → 3 _{1,3}	7	6.0	-6.0647	2.4	88	13.4	clean transition
46737.73	4 _{0,4} → 3 _{0,3}	7	3.4	-6.0064	2.6	122	8.6	clean transition
71800.18	6 _{1,5} → 5 _{1,4}	13	11.3	-5.4428	3.5	121	6.3	clean transition
81630.08	7 _{0,7} → 6 _{0,6}	15	11.8	-5.2587	3.4	153	8.2	slightly blended with CH ₃ CH ₂ CHO
83749.29	7 _{1,6} → 6 _{1,5}	15	14.8	-5.2342	3.4	116	11.8	slightly blended with CH ₃ COOH
91251.84	8 _{1,8} → 7 _{1,7}	17	18.1	-5.1166	1.7	104	11.2	slightly blended with (CH ₂ OH) ₂ and unidentified species
93505.09	8 _{2,7} → 7 _{2,6}	17	26.5	-5.1061	1.4	46	6.0	blended with CH ₃ ¹⁸ OH

Notes. The integrated signal ($\int T_A^* d\nu$) and rms noise level are also provided and used to calculate the signal-to-noise ratio (S/N) of the detections.

^a $J_{K'_a K'_c}'' \rightarrow J_{K''_a K''_c}'$ (where the double prime indicates the upper state).

^b S/N is calculated from the integrated signal and noise level $\sigma = \text{rms} \sqrt{\delta\nu \text{FWHM}}$, where $\delta\nu$ is the velocity resolution of the spectra.

Table 2

Physical Parameters of the Species Derived by LTE Analysis in MADCUBA

Molecular Formula	N ($\times 10^{13}$ cm ⁻²)	T_{ex} (K)	v_{LSR} (km s ⁻¹)	FWHM (km s ⁻¹)	Abundance ($\frac{N(X)}{N(\text{H}_2)} \times 10^{-10}$)
trans – HC(O)SH	1.6 ± 0.1	10 ± 1	69.0 ^a	21.0 ^a	1.2 ± 0.2
cis – HC(O)SH	≤0.3	10.0	69.0 ^a	21.0 ^a	≤0.2
g – C ₂ H ₅ SH	4 ± 2	10 ± 5	69.0 ^a	20.0 ^a	3 ± 1
a – C ₂ H ₅ SH	≤2.3	9.91	69.0 ^a	20.0 ^a	≤1.7
CH ₃ SH, $K_a = 0$	46.8 ± 0.5	8.5 ± 0.1	68.0 ± 0.1	21.2 ± 0.3	...
CH ₃ SH, $K_a = 1$	18.6 ± 0.7	14.9 ± 0.5	68.8 ± 0.3	22.0 ± 0.7	...
CH ₃ SH	65 ± 2	48 ± 5
t – HCOOH	20 ± 4	10 ± 2	68 ± 2	22 ± 5	15 ± 4

Note.

^a Value fixed in the fit.

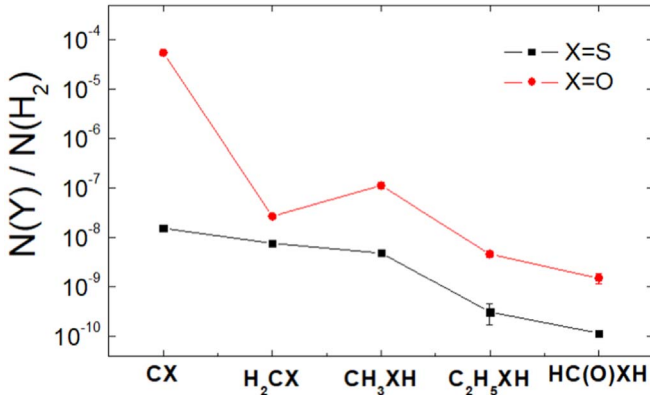


Figure 2. Molecular abundances with respect to H₂ for the reported detections and their OH analogs (CH₃OH, C₂H₅OH, and HC(O)OH) including CS, CO, H₂CO, and H₂CS. The lines that connect each dot do not have any physical meaning but are just a visual aid.

monosulfide (CS) and thioformaldehyde (H₂CS) with their O-bearing analogs, resulting in H₂CO/H₂CS = 3.5 and CO/CS = 3.5 × 10³ (Figure 2).

Since CH₃OH, C₂H₅OH, H₂CO, CO, and CS are optically thick toward G+0.693, the column density of these molecules was inferred from CH₃¹⁸OH, ¹³CH₃CH₂OH, H₂C¹⁸O, C¹⁸O, and ¹³C³⁴S. It was assumed ¹⁶O/¹⁸O = 250 and ¹²C/¹³C = 21

(Armijos-Abendaño et al. 2015) and ³²S/³⁴S = 22 (Wilson 1999). For HC(O)OH, the main isotopologue of HC(O)OH was employed as the ratio H¹²C(O)OH/H¹³C(O)OH, which is consistent with the ¹²C/¹³C ratio. For the sulfur analogs, except CS, we do not expect opacity issues due to their much lower abundances. The fit and transitions selected for these molecules are listed in Appendix B (Figures 5 and 6 and Tables 6, 7, 8, 9, and 10).

4. Discussion

4.1. Comparison with Previous Observations

We have compared different column density ratios obtained between the S-bearing compounds measured toward G+0.693 and two other sources, namely, Orion KL (Kolesniková et al. 2014) and Sgr B2(N2) (Müller et al. 2016; see Table 3).

From this table, we find that all abundance ratios measured in G+0.693 are strikingly similar. For instance, when we compare SH-bearing molecules with their OH analogs (CH₃OH/CH₃SH and C₂H₅OH/C₂H₅SH) we recover the trend already found in Figure 2, which might indicate a similar chemistry between OH- and SH-bearing species. The CH₃OH/CH₃SH ratio measured in G+0.693 is a factor of 5 lower than those found in Sgr B2(N2) and Orion KL, which indicates that G+0.693 is richer in sulfur-bearing species than the two massive hot cores. This may be related to the fact that the chemistry of this cloud is affected by large-scale shocks (Requena-Torres et al. 2006; Zeng et al. 2018).

Table 3
Relative Abundances of Molecules and Comparison with Other Sources

Source	CH ₃ SH/C ₂ H ₅ SH	CH ₃ OH/C ₂ H ₅ OH	CH ₃ OH/CH ₃ SH	C ₂ H ₅ OH/C ₂ H ₅ SH
G+0.693	16 ± 7	24 ± 4 ^{c,d}	23 ± 3 ^c	15 ± 7 ^d
Sgr B2(N2) ^a	≥21	20	118	≥125
Orion KL ^b	5	31	120	20

Notes.

^a Data taken from Müller et al. (2016).

^b Data taken from Kolesníková et al. (2014).

^c Data inferred from CH₃¹⁸OH assuming ¹⁶O/¹⁸O = 250 (Armijos-Abendaño et al. 2015).

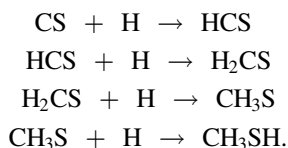
^d Data inferred from ¹³CH₃CH₂OH assuming ¹²C/¹³C = 21 (Armijos-Abendaño et al. 2015).

Since sulfur is heavily depleted on grains (possibly in the form of S₈ and other sulfur allotropes; Shingledecker et al. 2020), the sputtering of dust grains induced by shocks could liberate a significant fraction of the locked sulfur.

The C₂H₅OH/C₂H₅SH ratio is comparable to the value determined toward Orion KL while there is a difference by a factor of 10 between G+0.693 and the upper limit toward Sgr B2(N2). All these ratios suggest that sulfur is less incorporated into molecules than oxygen, which again might be due to the fact that a significant fraction of sulfur is locked up in grains.

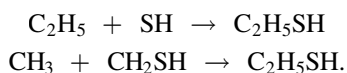
4.2. Interstellar Chemistry of Thiols

In order to understand the chemistry of S-bearing compounds, different models have been proposed based on the simple molecules detected in the gas phase (Müller et al. 2016; Gorai et al. 2017; Lamberts 2018; Laas & Caselli 2019). CH₃SH is thought to be formed on grain surfaces by sequential hydrogenations starting from CS:



The last step could also yield other products, but theoretical calculations show a branching ratio (br) of ~75% for CH₃SH (Lamberts 2018). This is a viable mechanism since approximately half of the CS present in the ices is available to undergo hydrogenation, while the other half is converted to OCS (Palumbo et al. 1997). Once formed, CH₃SH could remain stored in the ices until released by grain sputtering in the large-scale shocks present in G+0.693 (Requena-Torres et al. 2006; Zeng et al. 2020).

Likewise, ethyl mercaptan is proposed to be formed by radical-radical reactions (Müller et al. 2016; Gorai et al. 2017) such as



Observations of the reactants could give us a hint of the dominant reaction based on the measured column densities. However, chemical modeling is needed to understand the efficiency of these formation routes. Note that CH₂SH is the main product of the hydrogenation of H₂CS (Lamberts 2018).

To our knowledge, there is no information available in the literature about the chemistry of HC(O)SH. However, we can make a guess and assume similar astrochemical pathways for the SH-based species as the ones for their OH analogs. A possible pathway to HC(O)SH could mimic the formation of HC(O)OH. Ioppolo et al. (2011) showed that the formation of HC(O)OH starts from CO and the OH radical in the ice. The thiol-equivalent

reaction would be

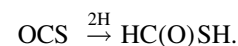


The first step has been found to be efficient by Adriaens et al. (2010), but further experimental and/or theoretical work is needed to investigate whether HSCO could be hydrogenated further.

Another possibility could be



or



The first reaction was initially proposed for HC(O)OH by Garrod & Herbst (2006) and could be a viable mechanism since SH can be formed on surfaces from S+H or H+H₂S via tunneling (Vidal et al. 2017). The second involves the sequential hydrogenation of OCS, which is a molecule detected in ices (Palumbo et al. 1997) and it is moderately abundant in the gas phase in G+0.693 ($N > 10^{15} \text{ cm}^{-2}$; Armijos-Abendaño et al. 2015). Theoretical studies about these chemical networks are currently under work and will be presented in a forthcoming paper (G. Molpeceres et al. 2021, in preparation).

4.3. Implications for Theories on the Origin of Life

As it was previously stated, thioacids and thioesters have been proposed as key agents in the polymerization of amino acids into peptides and proteins (Foden et al. 2020; Muchowska & Moran 2020). In some of these works, it is stressed the importance of cysteine, HSCH₂CH(NH₂)COOH, as the primary organic source of sulfide in biology and a key catalyst in the abiotic polymerization of peptides. Smaller thiols such as CH₃SH and C₂H₅SH are also believed to play a key role in prebiotic chemistry and in theories about the origin of life, since they are precursors for the synthesis of the amino acids methionine and ethionine (Parker et al. 2011). In turn, HC(O)SH could be an important ingredient in the phosphorylation of nucleosides as demonstrated by Lohrmann & Orgel (1968).

The idea of an extraterrestrial delivery of SH compounds onto Earth, is supported by the detection of CH₃SH in Murchinson carbonaceous chondrite (Tingle et al. 1991) and in the coma of the 67P/Churyumov–Gerasimenko comet (Calmonte et al. 2016). In the latter, C₂H₆S was also detected (either in the form of C₂H₅SH or CH₃SCH₃, dimethyl sulfide) with an abundance ratio CH₃SH/C₂H₆S ~ 10 that is consistent with the value obtained in G+0.693 (Table 3). This resemblance could be possibly indicate a presolar origin of these compounds. Still, further studies are required within other regions in the ISM and planetary bodies to make a proper connection.

In summary, not only have our observations confirmed the presence of sulfur-bearing complex organics such as CH_3SH and $\text{C}_2\text{H}_5\text{SH}$, but they also have revealed the existence of the simplest thioacid known, HC(O)SH , in the ISM.

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Facilities: IRAM 30 m, Yebes 40 m.

Software: MADCUBA.

Appendix A Detections of $\text{C}_2\text{H}_5\text{SH}$ and CH_3SH in G+0.693

In Figures 3 and 4 and Tables 4 and 5 the fit of $g\text{-C}_2\text{H}_5\text{SH}$ and CH_3SH is shown. The fitting procedure followed has been the same as $t\text{-HC(O)SH}$, explained in the main text.

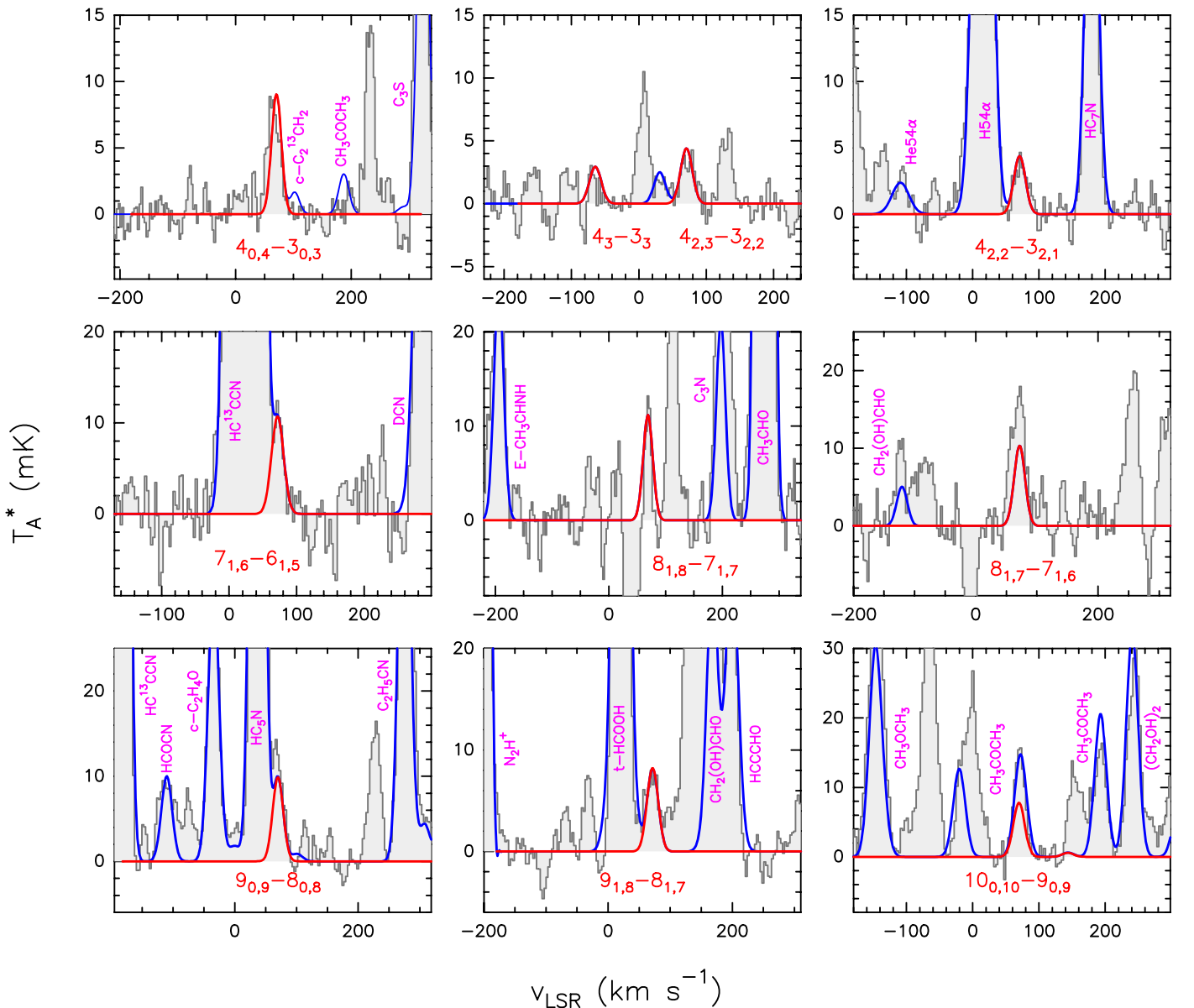


Figure 3. Cleanest and brightest lines of $g\text{-C}_2\text{H}_5\text{SH}$ detected toward G+0.693. The red line shows the best LTE fit to the observed spectra (represented by the black lines) with their corresponding quantum numbers in red (J_{K_a,K_c} ; see Table 4 for a description of their tunneling states). The data have been smoothed up to 3 km s^{-1} for an optimal line visualization. The blue lines show the spectra including the emission of all the molecules searched toward the cloud. Note that these lines are tagged with their corresponding molecular compound in pink.

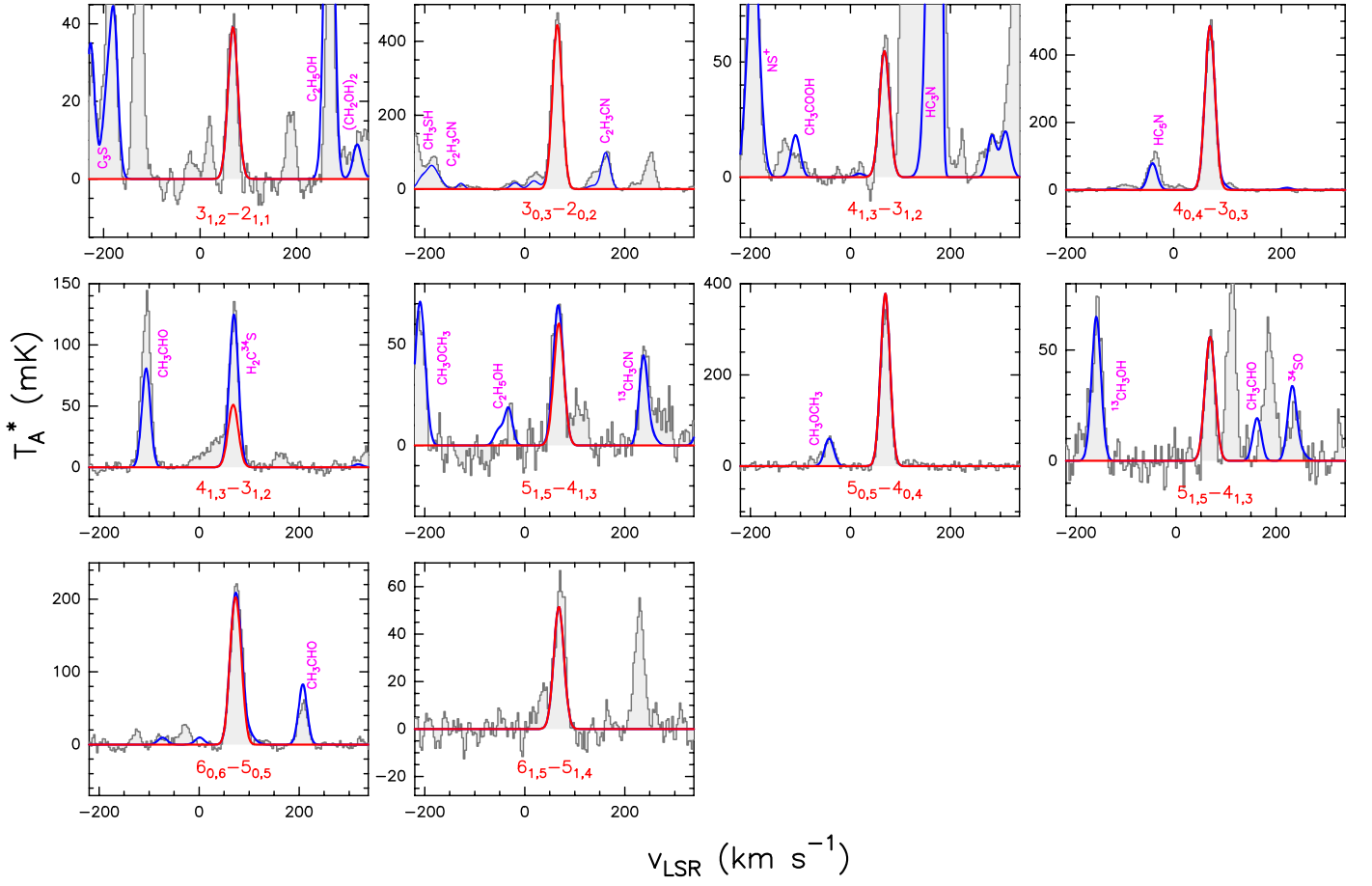


Figure 4. Cleanest and brightest lines of CH_3SH detected toward G+0.693. The red line shows the best LTE fit to the observed spectra (represented by the black lines) with their corresponding quantum numbers in red (J_{K_a, K_c} ; see Table 5 for the description of their torsional states). The data have been smoothed up to 3 km s^{-1} for an optimal line visualization. The blue lines show the spectra including the emission of all the molecules searched toward the cloud. Note that these lines are tagged with their corresponding molecular compound in pink.

Table 4
Lines of g- $\text{C}_2\text{H}_5\text{SH}$ Detected toward G+0.693 with Their Corresponding QNs and Tunneling States (TSs), $\log A_{ul}$, g_u , and E_u

Rest Frequency (MHz)	QNs	TS $\nu'' \rightarrow \nu'$	g_u	E_u (K)	$\log A_{ul}$ (s^{-1})	rms (mK)	$\int T_A^* d\nu$ (mK km s^{-1})	S/N	Comments
40499.173	$4_{0,4} \rightarrow 3_{0,3}$	$1 \rightarrow 1$	9	3.0	-6.1161	1.4	225	29	clean transition ^a
40499.591	$4_{0,4} \rightarrow 3_{0,3}$	$0 \rightarrow 0$	9	2.9	-6.1163	1.4
40558.849	$4_{2,3} \rightarrow 3_{2,2}$	$1 \rightarrow 1$	9	7.6	-6.2391	1.2	107	16	clean transition ^a
40559.284	$4_{2,3} \rightarrow 3_{2,2}$	$0 \rightarrow 0$	9	7.5	-6.2392	1.2
40576.963	$4_{3,2} \rightarrow 3_{3,1}$	$1 \rightarrow 1$	9	13.2	-6.2004	1.2	72	11	clean transition ^a
40577.299	$4_{3,1} \rightarrow 3_{3,0}$	$1 \rightarrow 1$	9	13.2	-6.2004	1.2
40577.432	$4_{3,2} \rightarrow 3_{3,1}$	$0 \rightarrow 0$	9	13.2	-6.2004	1.2
40577.769	$4_{3,1} \rightarrow 3_{3,0}$	$0 \rightarrow 0$	9	13.2	-6.2004	1.2
40622.316	$4_{2,2} \rightarrow 3_{2,1}$	$1 \rightarrow 1$	9	7.6	-6.2370	1.2	107	16	clean transition ^a
40622.792	$4_{2,2} \rightarrow 3_{2,1}$	$0 \rightarrow 0$	9	7.5	-6.2371	1.2
72466.861	$7_{1,6} \rightarrow 6_{1,5}$	$1 \rightarrow 1$	15	11.7	-5.3458	2.8	261	17	clean transition ^a
72468.152	$7_{1,6} \rightarrow 6_{1,5}$	$0 \rightarrow 0$	15	11.6	-5.3459	2.8
79204.501	$8_{1,8} \rightarrow 7_{1,7}$	$1 \rightarrow 1$	17	14.6	-5.2243	1.8	204	24	clean transition ^a
79204.501	$8_{1,8} \rightarrow 7_{1,7}$	$0 \rightarrow 0$	17	14.6	-5.2243	1.8
82782.441	$8_{1,7} \rightarrow 7_{1,6}$	$1 \rightarrow 1$	17	15.2	-5.1667	1.8	233	30	slightly blended with unidentified line ^a
82783.912	$8_{1,7} \rightarrow 7_{1,6}$	$0 \rightarrow 0$	17	15.1	-5.1668	1.4
90516.341	$9_{0,9} \rightarrow 8_{0,8}$	$1 \rightarrow 1$	19	17.6	-5.0412	1.2	223	34	slightly blended with unidentified line ^a
90516.926	$9_{0,9} \rightarrow 8_{0,8}$	$0 \rightarrow 0$	19	17.5	-5.0414	1.2
93082.083	$9_{1,8} \rightarrow 8_{1,7}$	$1 \rightarrow 1$	19	19.1	-5.0096	2.0	191	18	clean transition ^a
93083.720	$9_{1,8} \rightarrow 8_{1,7}$	$0 \rightarrow 0$	19	19.1	-5.0097	2.0

Table 4
(Continued)

Rest Frequency (MHz)	QNs	TS $\nu'' \rightarrow \nu'$	g_u	E_u (K)	$\log A_{ul}$ (s^{-1})	rms (mK)	$\int T_A^* d\nu$ (mK km s $^{-1}$)	S/N	Comments
100391.240	$10_{0,10} \rightarrow 9_{0,9}$	$1 \rightarrow 1$	21	21.9	-4.9042	1.8	171	17	blended with $CH_3COCH_3^a$
100391.791	$10_{0,10} \rightarrow 9_{0,9}$	$0 \rightarrow 0$	21	21.8	-4.9042	1.8

Notes. $\int T_A^* d\nu$ and rms are also provided and used to calculate the S/N of the detected lines.

^a In these cases, the lines are composed by a blend of individual g-C₂H₅SH lines. Here, S/N is calculated according to Figure 3.

Table 5

Lines of CH₃SH Detected toward G+0.693 with Their Corresponding Quantum Numbers (QNs) and Torsional Symmetry (T_s), Logarithm of the Einstein Coefficients ($\log A_{ul}$), Degeneracy (g_u), and Energy (E_u) of the Upper State

Rest Frequency (MHz)	QNs	T_s	g_u	E_u (K)	$\log A_{ul}$ (s^{-1})	rms (mK)	I_{peak} (mK)	S/N	Comments
75085.898	$3_{1,2} \rightarrow 2_{1,1}$	A ⁺	7	8.7	-5.5086	2.7	39	14	clean transition
75862.889	$3_{0,3} \rightarrow 2_{0,2}$	A ⁺	7	3.6	-5.4444	2.3	260	113	clean transition
75864.422	$3_{0,3} \rightarrow 2_{0,2}$	E	7	5.1	-5.4443	2.3	221	96	clean transition
100110.219	$4_{1,3} \rightarrow 3_{1,2}$	A ⁺	9	12.3	-5.0966	3.1	55	18	clean transition
101139.150	$4_{0,4} \rightarrow 3_{0,3}$	A ⁺	9	7.3	-5.0554	2.9	271	93	clean transition
101139.655	$4_{0,4} \rightarrow 3_{0,3}$	E	9	8.7	-5.0543	2.9	230	79	clean transition
101284.366	$4_{1,3} \rightarrow 3_{1,2}$	E	9	13.5	-5.0818	2.9	51	18	blended with H ₂ C ³⁴ S
125130.863	$5_{1,5} \rightarrow 4_{1,3}$	A ⁺	11	17.1	-4.7869	6.2	61	10	slightly blended with HC ₅ N
126403.834	$5_{0,5} \rightarrow 4_{0,4}$	E	11	13.6	-4.7556	6.4	183	29	clean transition
126405.676	$5_{0,5} \rightarrow 4_{0,4}$	A ⁺	11	12.2	-4.7556	6.4	215	34	clean transition
126683.419	$5_{1,5} \rightarrow 4_{1,9}$	A ⁺	11	18.4	-4.7707	6.5	56	9	clean transition
151654.218	$6_{0,6} \rightarrow 5_{0,5}$	E	13	19.6	-4.5123	5.4	116	22	clean transition
151660.047	$6_{0,6} \rightarrow 5_{0,5}$	A ⁺	13	18.2	-4.5123	5.4	136	25	clean transition
152129.018	$6_{1,5} \rightarrow 5_{1,4}$	E	13	24.4	-4.5210	4.5	52	12	clean transition

Note. The peak intensity (I_{peak}) and rms noise level are also provided and used to calculate the signal-to-noise ratio (S/N) of the detected lines.

Appendix B

Figures and Tables of $^{13}\text{C}^{34}\text{S}$, H_2CS , $\text{H}_2\text{C}^{18}\text{O}$, $\text{CH}_3^{18}\text{OH}$, and $^{13}\text{CH}_3\text{CH}_2\text{OH}$

We show a few of the intense lines selected within the targeted frequency coverage in G+0.693 for some of the species used in the molecular abundances calculations in Figure 2. Note that other visible lines not shown in these figures or tables are either heavily blended or in a bad part of the spectra.

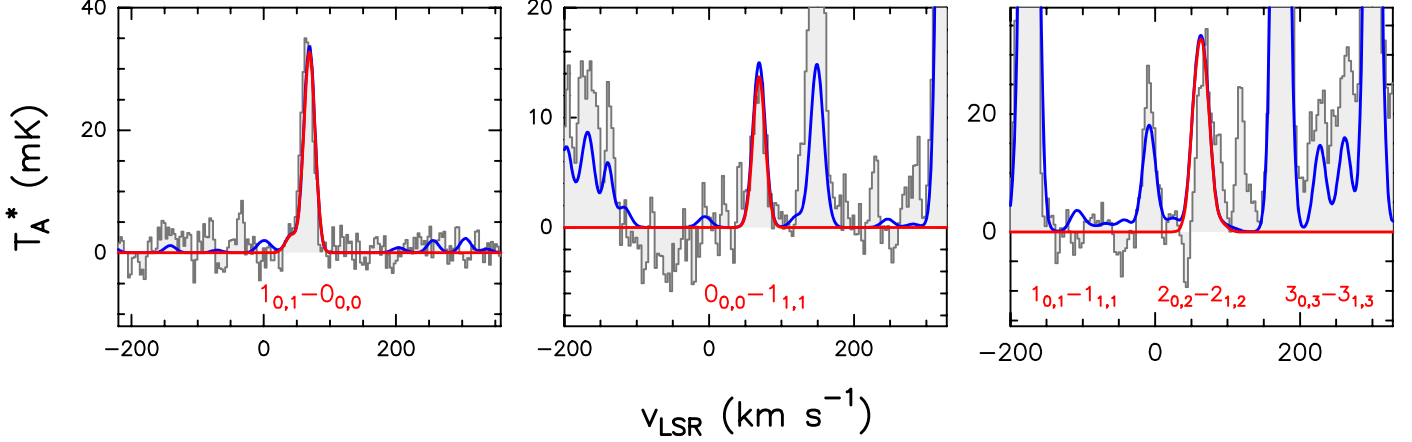


Figure 5. Selected lines of $\text{CH}_3^{18}\text{OH}$ detected toward G+0.693. Black, red, and blue lines represent the observational data, the individual lines of $\text{CH}_3^{18}\text{OH}$, and the global fit considering all the species detected, respectively. See Table 9 for a complete description of their quantum numbers).

Table 6
Lines of $^{13}\text{C}^{34}\text{S}$ Detected toward G+0.693 with Their Corresponding Spectroscopic Information and Parameters Derived from the LTE Fit

Rest Frequency (MHz)	QNs	g_u	E_u (K)	$\log A_{ul}$ (s^{-1})	rms (mK)	$\int T_A^* d\nu$ (mK km s^{-1})	S/N	Comments
45463.424	$1 \rightarrow 0$	3	0.0	-5.8546	2.0	364	>20	clean transition
90926.026	$2 \rightarrow 1$	5	2.0	-4.8723	2.5	854	>20	clean transition
136387.028	$3 \rightarrow 2$	7	6.6	-4.3141	2.1	796	>20	clean transition

Table 7
Lines of H_2CS Detected toward G+0.693 with $I_{\text{peak}} > 0.8\text{K}$ and Their Corresponding Spectroscopic Information and Parameters Derived from the LTE Fit

Rest Frequency (MHz)	QNs	g_u	E_u (K)	$\log A_{ul}$ (s^{-1})	rms (mK)	I_{peak} (K)	S/N	Comments
101477.8048	$3_{1,3} \rightarrow 2_{1,2}$	21	2.0	-4.8996	2.9	1.213	>20	clean transition
103040.447	$3_{0,3} \rightarrow 2_{0,2}$	7	5.0	-4.8285	2.6	0.951	>20	clean transition
104617.027	$3_{1,2} \rightarrow 2_{1,1}$	27	18.2	-4.8599	2.1	1.169	>20	clean transition
135298.26	$4_{1,4} \rightarrow 3_{1,3}$	27	22.9	-4.4859	7.0	1.213	>20	clean transition
137371.21	$4_{0,4} \rightarrow 3_{0,3}$	9	9.9	-4.438	3.1	0.823	>20	clean transition
139483.68	$4_{1,3} \rightarrow 3_{1,2}$	27	23.2	-4.4462	2.7	1.197	>20	clean transition

Table 8
Lines of $\text{H}_2\text{C}^{18}\text{O}$ Detected toward G+0.693 with Their Corresponding Spectroscopic Information and Parameters Derived from the LTE Fit

Rest Frequency (MHz)	QNs	g_u	E_u (K)	$\log A_{ul}$ (s^{-1})	rms (mK)	$\int T_A^* d\nu$ (mK km s^{-1})	S/N	Comments
134435.920	$2_{1,2} \rightarrow 1_{1,1}$	15	2.0	-4.3361	6.7	839	>20	slightly blended with CH_3CHO
138770.86	$2_{0,2} \rightarrow 1_{0,1}$	5	3.3	-4.1697	2.7	814	>20	clean transition
143213.07	$2_{1,1} \rightarrow 1_{1,0}$	15	15.3	-4.2536	2.1	824	>20	clean transition

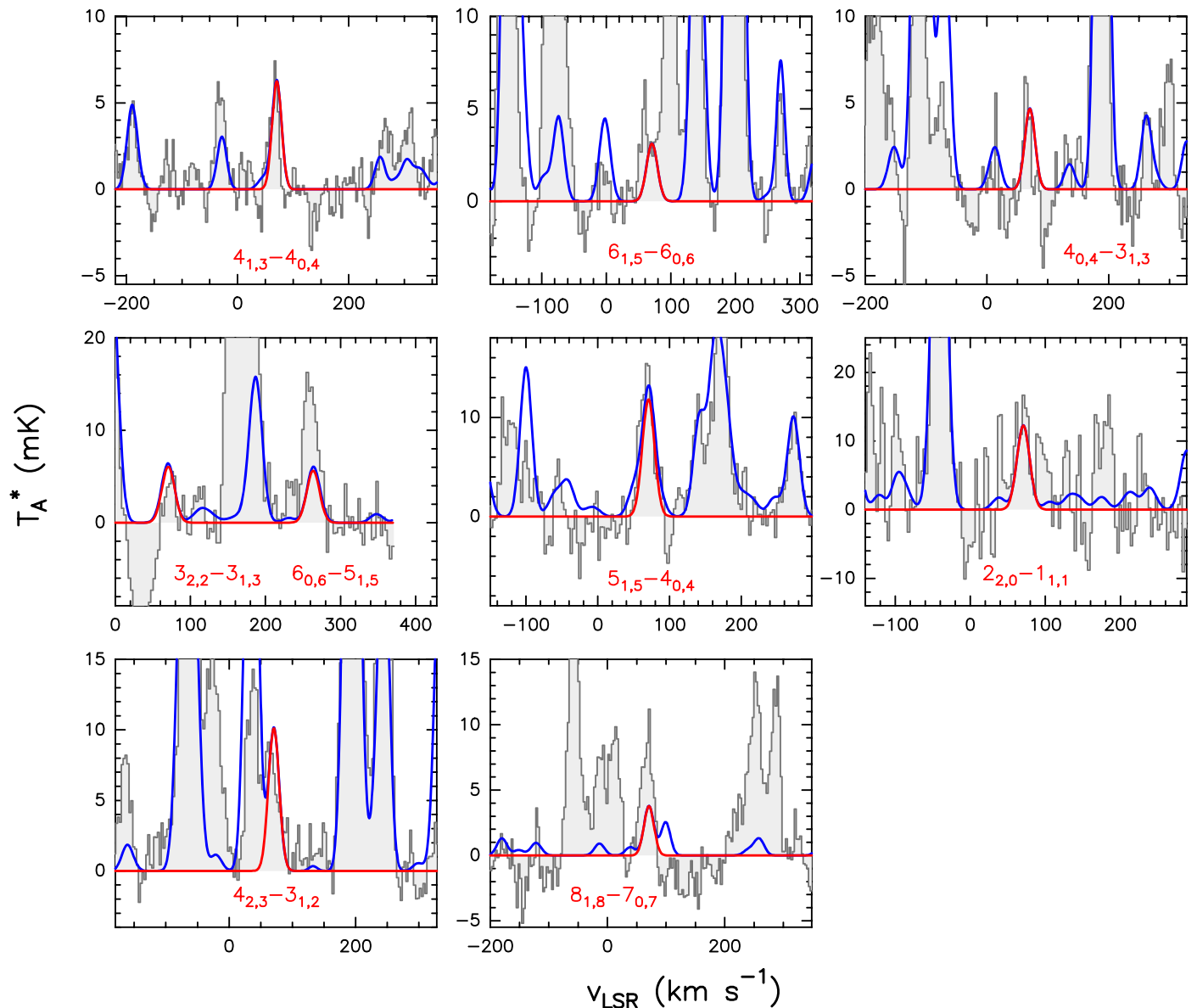


Figure 6. Selected lines of anti- $^{13}\text{CH}_3\text{CH}_2\text{OH}$ detected toward G+0.693. Black, red, and blue lines represent the observational data, the individual lines of $^{13}\text{CH}_3\text{CH}_2\text{OH}$, and the global fit with all species, respectively.

Table 9
Lines of CH_3^8OH Detected toward G+0.693 with Their Corresponding Spectroscopic Information and Parameters Derived from the LTE Fit

Rest Frequency (MHz)	QNs	T_s	g_u	E_u (K)	$\log A_{ul}$ (s^{-1})	rms (mK)	$\int T_A^* d\nu$ (mK km s^{-1})	S/N	Comments
46364.313	$1_{0,1} \rightarrow 0_{0,0}$	A	12	0.0	-4.8985	2.8	709	>20	clean transition
93505.902	$2_{1,1} \rightarrow 1_{1,0}$	A	20	16.6	-5.6353	1.9	68	7	blended with t-HC(O)SH
150698.06	$1_{0,1} \rightarrow 1_{1,1}$	E	12	7.8	-4.7223	3.1	466	>20	clean transition
150704.121	$3_{0,3} \rightarrow 3_{1,3}$	E	28	19.0	-4.7330	3.1	127	8	clean transition
150704.173	$2_{0,2} \rightarrow 2_{1,2}$	E	20	12.3	-4.7266	3.1	328	19	clean transition

Table 10
Lines of anti- $^{13}\text{CH}_3\text{CH}_2\text{OH}$ Detected toward G+0.693 with Their Corresponding Spectroscopic Information and Parameters Derived from the LTE Fit

Rest Frequency (MHz)	QNs	g_u	E_u (K)	$\log A_{ul}$ (s^{-1})	rms (mK)	$\int T_A^* d\nu$ (mK km s^{-1})	S/N	Comments
32509.32	$4_{1,3} \rightarrow 4_{0,4}$	9	8.2	-6.4293	1.3	134	19	clean transition
40435.51	$6_{1,5} \rightarrow 6_{0,6}$	13	17.0	-6.2050	1.5	67	8	slightly blended with unidentified transition
44748.194	$4_{0,4} \rightarrow 3_{1,3}$	9	6.0	-6.3778	2.1	100	9	clean transition
82170.297	$6_{0,6} \rightarrow 5_{1,5}$	13	13.1	-5.4772	3.5	121	6	blended with unidentified transition
82223.219	$3_{2,2} \rightarrow 3_{1,3}$	7	6.0	-5.5620	2.8	130	9	clean transition
102934.258	$5_{1,5} \rightarrow 4_{0,4}$	11	8.2	-5.1178	2.6	253	18	clean transition
113369.368	$2_{2,0} \rightarrow 1_{1,1}$	5	2.1	-4.9878	6.8	262	7	clean transition
143297.176	$4_{2,3} \rightarrow 3_{1,2}$	9	6.3	-4.8286	2.2	216	18	clean transition
144465.114	$8_{1,8} \rightarrow 7_{0,7}$	17	22.7	-4.6339	1.7	81	9	blended with unidentified transition

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